

Study of TiO₂ Nanotube-based Photocatalysts

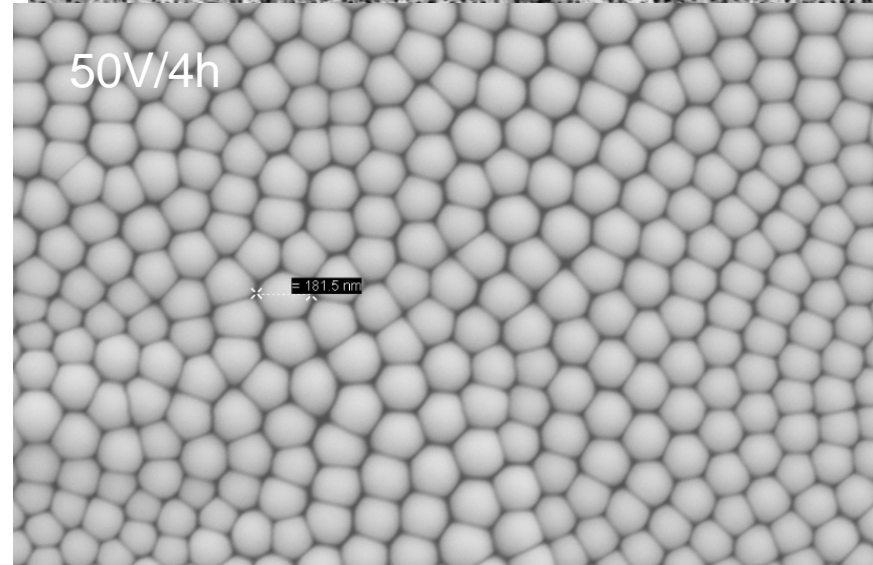
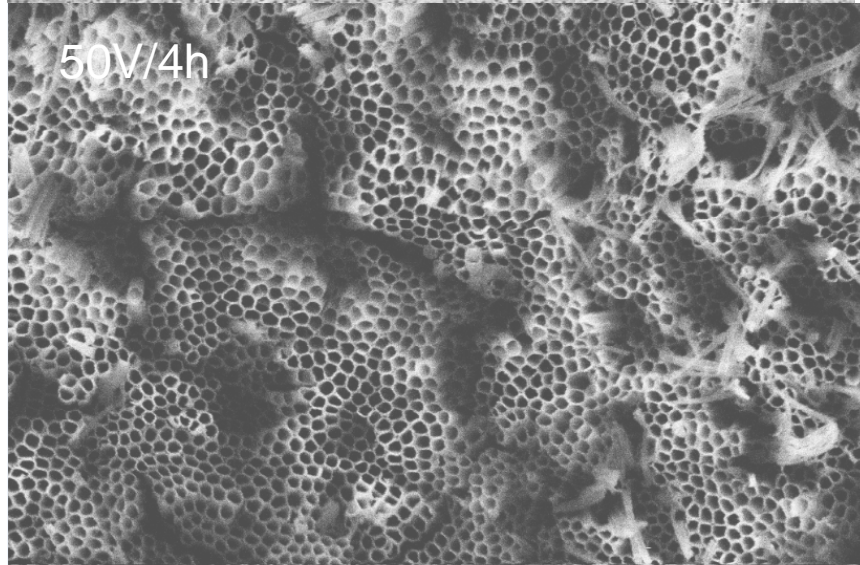
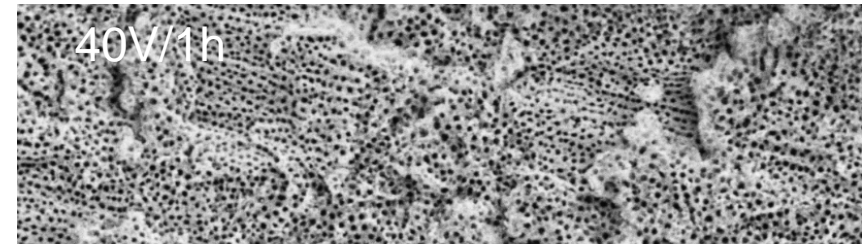
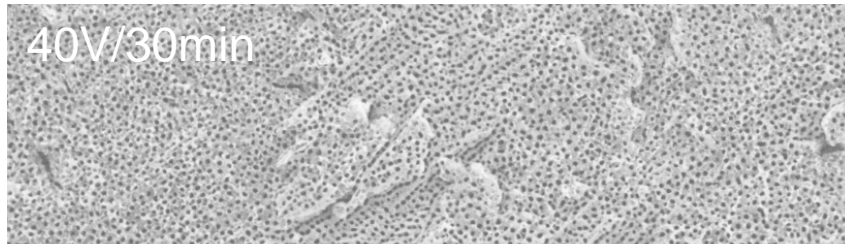
—by Jun Li

1. Formation of ordered TiO₂ NTs array
2. Quantum dots sensitized TiO₂ NTs
3. Nanotube-based newly TiO₂ NP
4. TiO₂ NTs template for Nanowire growth

Electrolyte: 0.3wt%NH₄F+2vol%H₂O+EG (80ml)

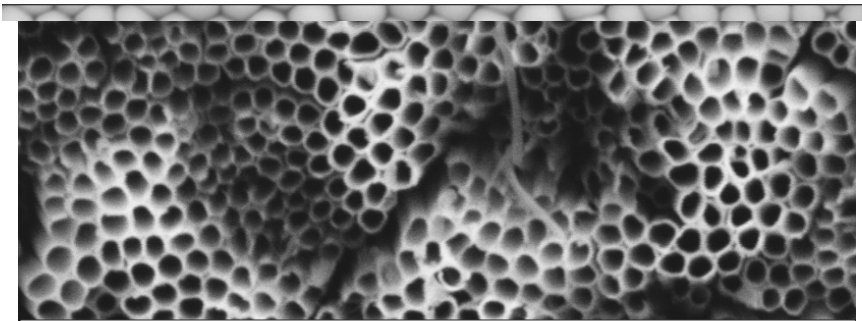
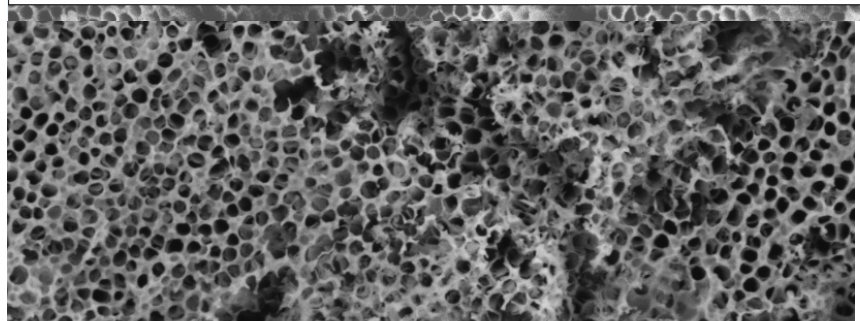
A two-step electrochemical anodization method:

Ti foil was first anodized at 50V for 4h, then ultrasonically rinsed in 1M HCl to peel off the first layer. Subsequently, a second anodization of Ti substrate was performed at 50V for 1h



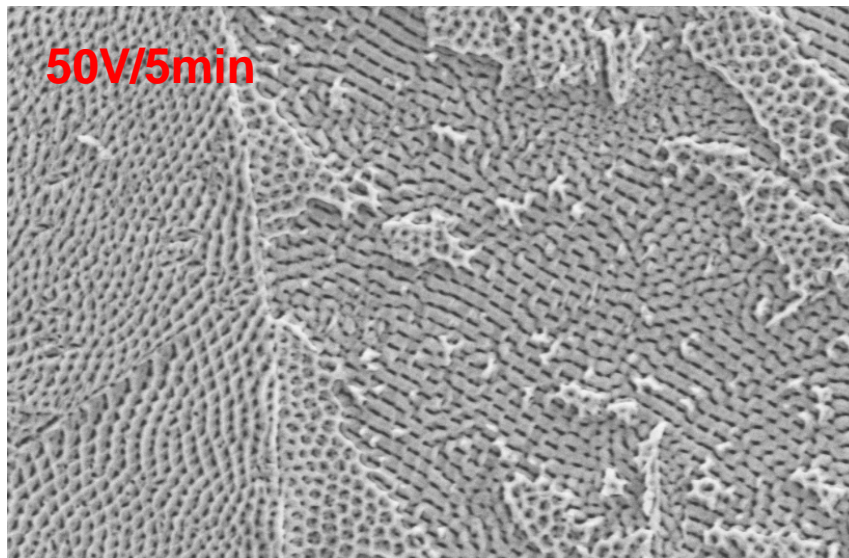
Mag = 17.26 K X | 1 μ m | EHT = 3.00 kV | Signal A = SE2 | Date :19 Mar 2013
WD = 6 mm | Photo No. = 2583 | Time :10:56:50

Mag = 41.98 K X | 200nm | EHT = 3.00 kV | Signal A = SE2 | Date :19 Mar 2013
WD = 6 mm | Photo No. = 2588 | Time :11:14:33

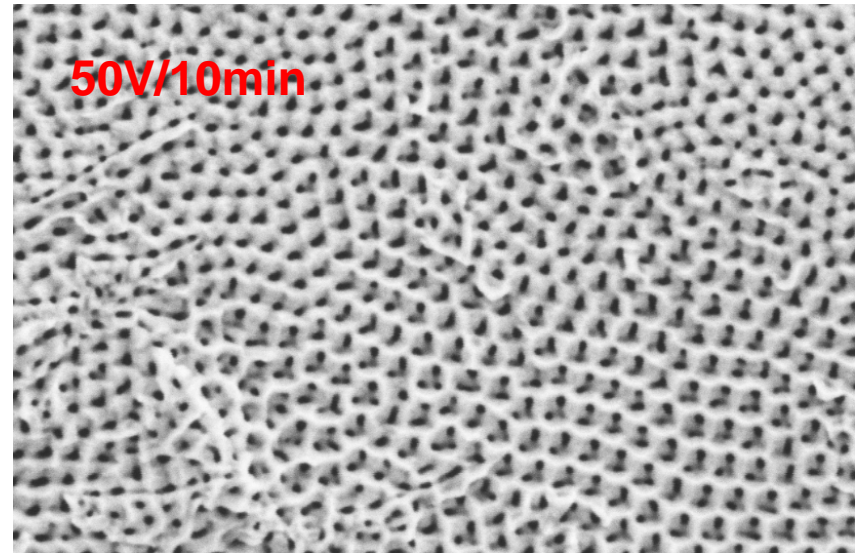


Mag = 25.00 K X | 200nm | EHT = 3.00 kV | Signal A = SE2 | Date :28 Mar 2013
WD = 6 mm | Time :14:17:59

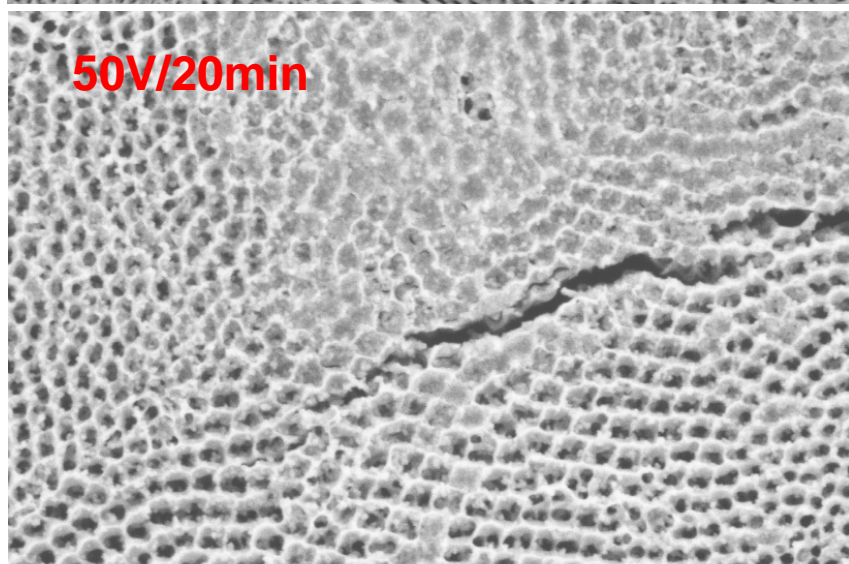
Mag = 35.11 K X | 200nm | EHT = 3.00 kV | Signal A = SE2 | Date :28 Mar 2013
WD = 6 mm | Time :15:09:51



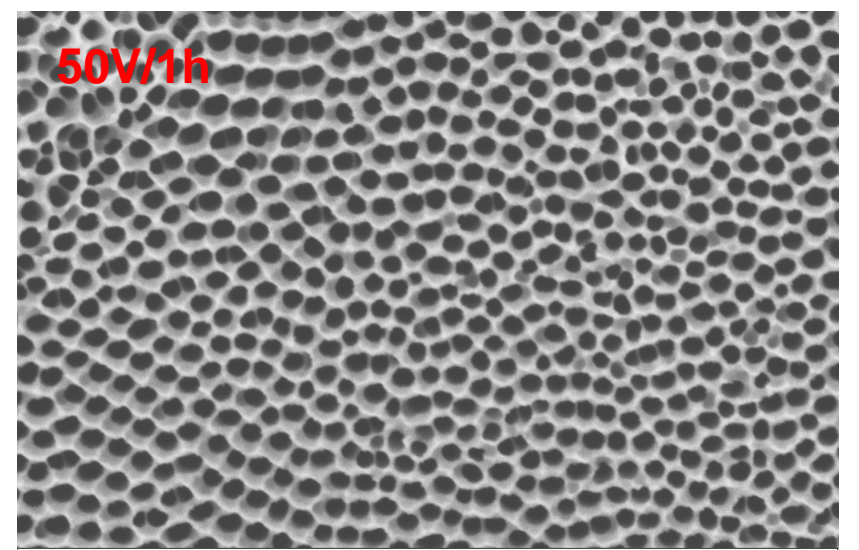
Mag = 15.51 K X 1 μm EHT = 3.00 kV Signal A = SE2 Date : 4 Apr 2013
 WD = 6 mm Time : 10:50:23



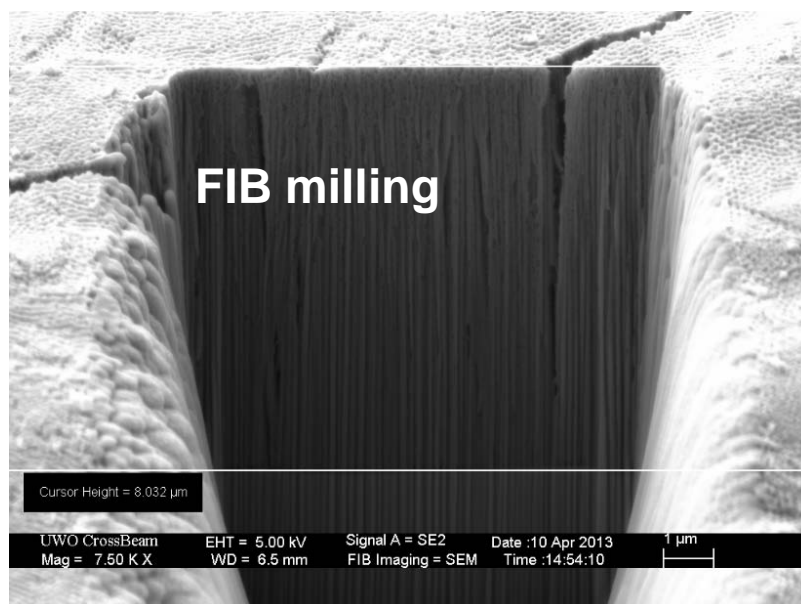
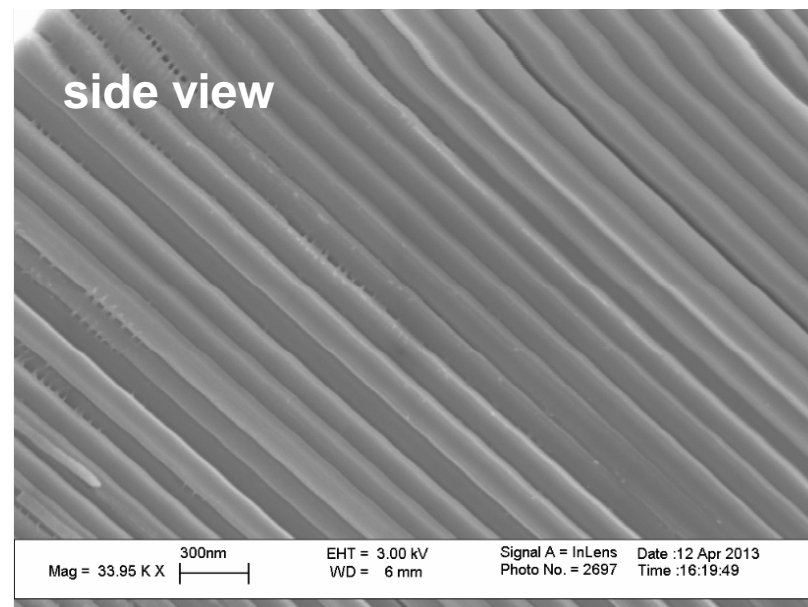
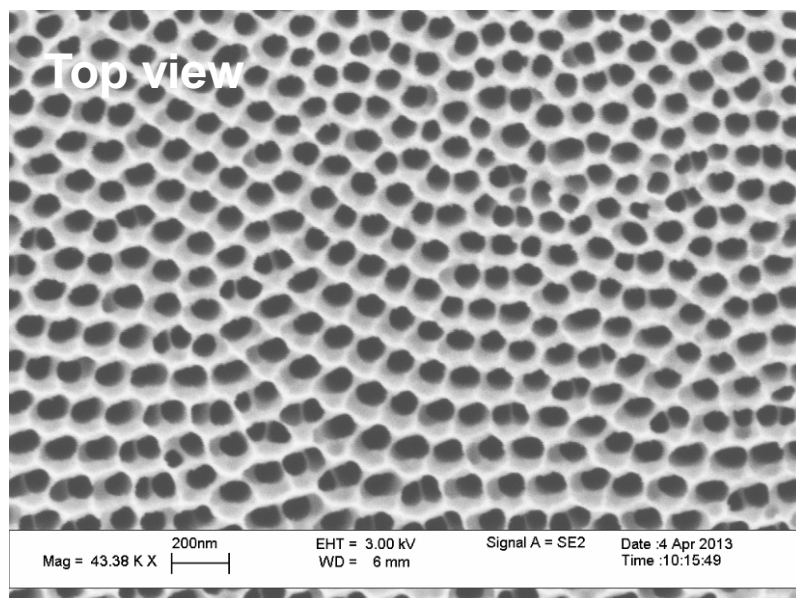
Mag = 30.10 K X 300nm EHT = 3.00 kV Signal A = SE2 Date : 4 Apr 2013
 WD = 6 mm Time : 10:34:59



Mag = 27.50 K X 200nm EHT = 3.00 kV Signal A = SE2 Date : 4 Apr 2013
 WD = 6 mm Time : 10:28:57



Mag = 34.15 K X 200nm EHT = 3.00 kV Signal A = SE2 Date : 4 Apr 2013
 WD = 6 mm Time : 10:13:12



Average growth rate:134nm/min

Fermi level: Pd < TiO₂

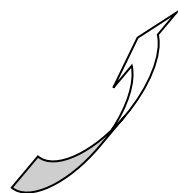
photoexcited

e⁻TiO₂ → Pd particles

reducing electron-hole recombination
 efficient charge separation
 greater photocatalytic reactions



What's the best annealing temperature for highly ordered TiO₂ NTs which is best for Pd coating in order to reduce electron-hole recombination and get a better photocatalytic property



Different annealing temperature leads to different crystal composition of TiO₂

P25: 25%anatase+75%rutile

High photocatalytic property of P25 is due to the mixed-phase TiO₂, which provides unique charge transfer pathways



Hydrothermal method: orderly TiO₂ NTs+PVP+NaI+PdCl₂

Before, top view

Next step:

Find the best composition for Pd QDs sensitization to obtain Pd-TiO₂ NTs

Different annealing temperature for as-prepared Pd-TiO₂ NTs to obtain

different Pd/TiO₂ NTAs (Annealing TiO₂ Nanotube Arrays)

After, top view

Mag = 102.46 K X EHT = 3.00 kV WD = 6 mm Signal A = SE2 Date :12 Apr 2013 Time :16:13:18

Mag = 166.89 K X EHT = 3.00 kV WD = 6 mm Signal A = InLens Photo No. = 2686 Date :12 Apr 2013 Time :15:45:59

Before, side view

After, side view

Mag = 46.70 K X EHT = 3.00 kV WD = 6 mm Signal A = InLens Photo No. = 2695 Date :12 Apr 2013 Time :16:17:43

Mag = 36.51 K X EHT = 3.00 kV WD = 6 mm Signal A = InLens Photo No. = 2693 Date :12 Apr 2013 Time :16:13:53

Correlation with **phase** and **electronic structure** study of Pd QDs sensitized TiO₂ NTs

SEM and Raman spectroscopy will be used to concentrate on the effect of Pd QDs for the phase transformation of TiO₂ NTs

HRTEM will be used to focus on the loading amount of Pd QDs on the surface of TiO₂ NTs and obtain the particle size distribution

Phase transformation and the light-emitting properties induced by calcination temperatures will be tracked using XANES and XEOL

XANES study:

Ti L_{3,2}-edge and O K-edge comparison between annealing-TiO₂ NTs arrays(ATNTA) and Pd-ATNTA will be used to concentrate on the effect of Pd QDs for the structure and bonding of TiO₂ NTs

Ti L_{3,2}-edge and O K-edge of Pd-ATNTA obtained from different annealing temperature will be used to distinguish different crystal phases among amorphous, anatase and rutile, and concentrate on the effect of Pd QDs for the phase transformation of TiO₂ NTs

Pd L-edge comparison between Pd-ATNTA and pure Pd will be used for the study of Pd electronic structure within Pd-ATNTA?

XEOL study:

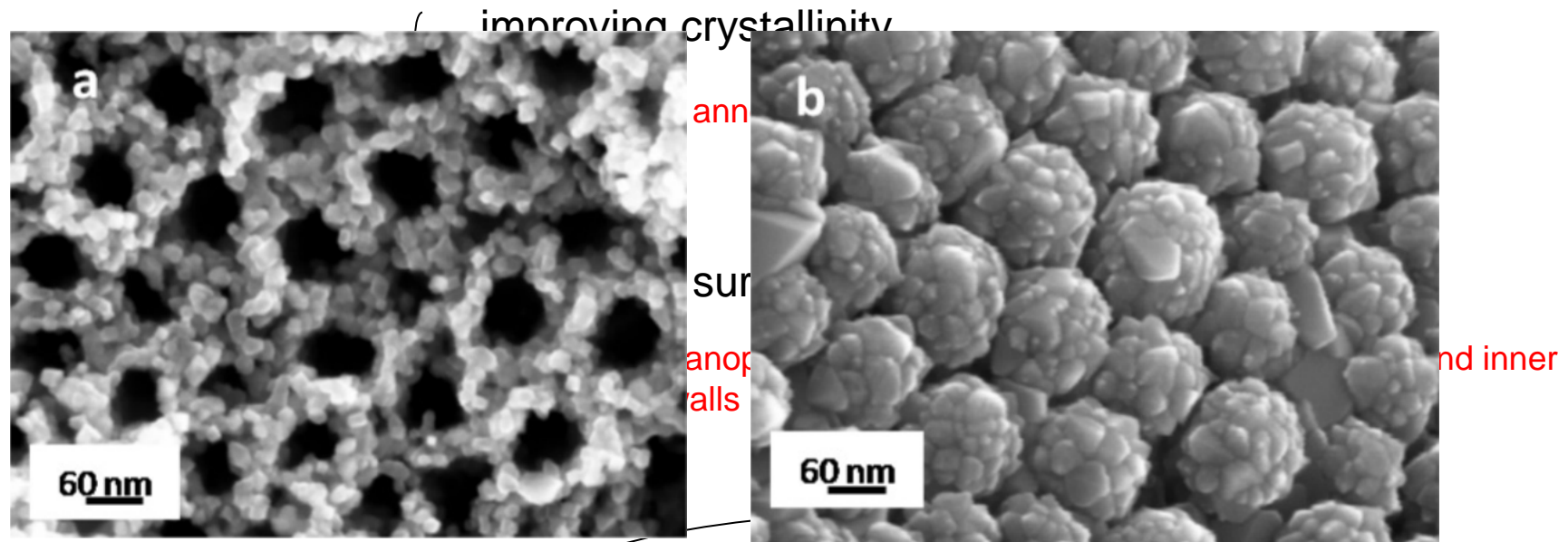
XEOL comparison between ATNTA and Pd-ATNTA will be used for the optical luminescence study and for the effect of Pd QDs sensitization

XEOL comparison among Pd-ATNTA samples obtained from different annealing temperature will be used for optical luminescence study

XEOL will be used to track XANES using photoluminescence yield(PLY/XANES) to reveal the element of the site among Pd-ATNTA samples obtained from different annealing temperature that is responsible for the luminescence

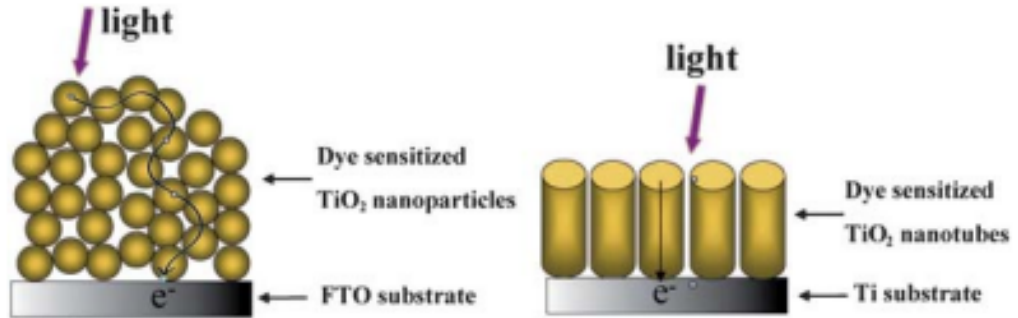
11 Nanotube-based newly TiO₂ NP(part three)

Electronic Structure of TiO₂ Nanotube-based Photocatalysts



Transformation from TiO₂ nanotube to nanoparticles

NPs-NTs transformation to obtain NTs-based newly TiO₂ NP



A comparison of the electron pathways through nanoparticle and nanotubular structured TiO₂.

The anodized tube structures are expected to greatly improve the charge collection efficiency due to directional electron conduction with minor charge recombination in comparison to TiO₂ nanoparticle films

Then what kind of new properties can we get from this NTs-based newly TiO₂ NP with a much higher surface area

13 Nanotube-based newly TiO₂ NP(part three)

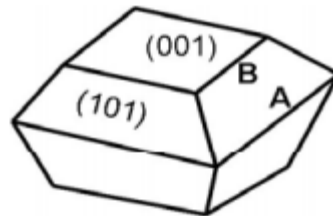
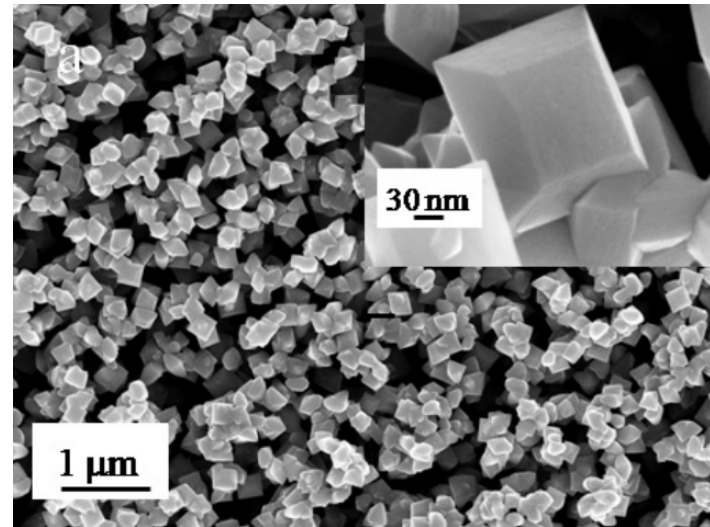
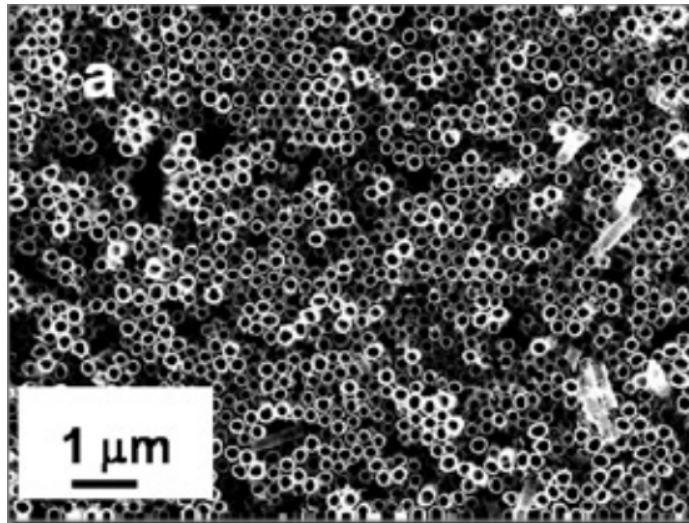
Electronic Structure of TiO₂ Nanotube-based Photocatalysts

Basic Outline:

NH₄F is used in the electrolyte for orderly TiO₂ NTs growth

Fluoride residue in the obtained NTs can change the morphology of TiO₂ NTs during a sealed annealing system

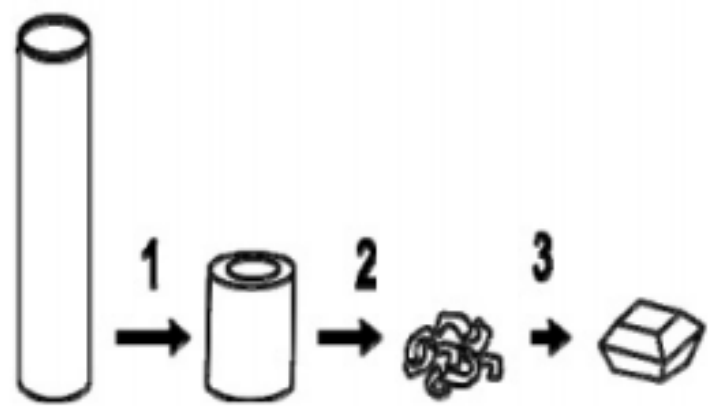
eg. Thermally annealing on a glass slide at 500°C for 30min with a ramping rate 16°C/min, the transformation happens:



Truncated tetragonal bipyramidal shape NP

Transformation mechanism

Lower F⁻ residue concentration



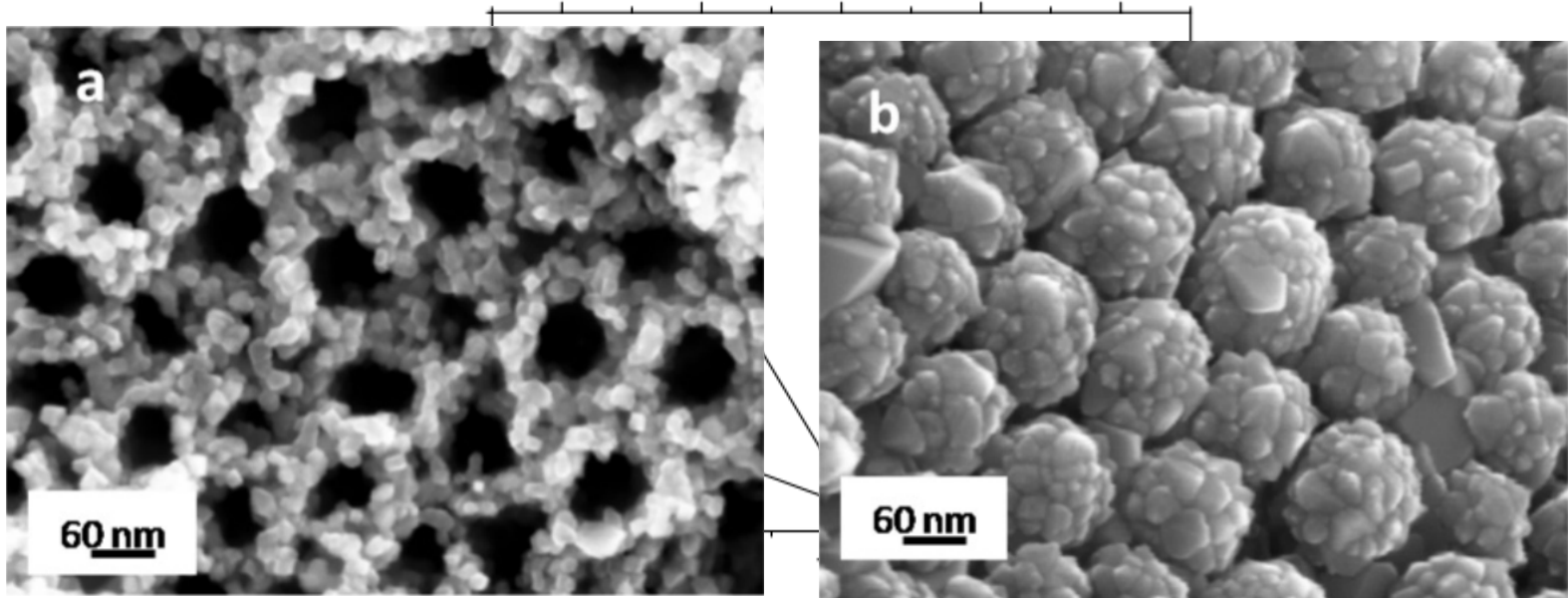
- (1) NT contraction
- (2) breaking of the contracted NT into small parts
- (3) merging of the broken parts and crystallizing into truncated pyramid shape nanoparticles

Higher F⁻ residue concentration

After contraction, the top of the thick NT walls crystallized directly to NPs before breaking down, without breaking the tubular geometry

Thus NTs-based newly TiO₂ NPs can be obtained from sealed annealing with a higher F⁻ residue concentration

Control growth: the most important part is the initial concentration of NH₄F in electrolyte



Top view (a) and bottom view (b) of selected areas of TiO₂ NT samples showing initial stages of the NT-NP transformation after 1 min annealing. The NH₄F concentration was 2% in these experiments. on a glass slide. Inset: a sketch of the truncated bipyramidal shape TiO₂ NPs showing (100) and (110) facets. The growth parameters as illustrated in the figure.

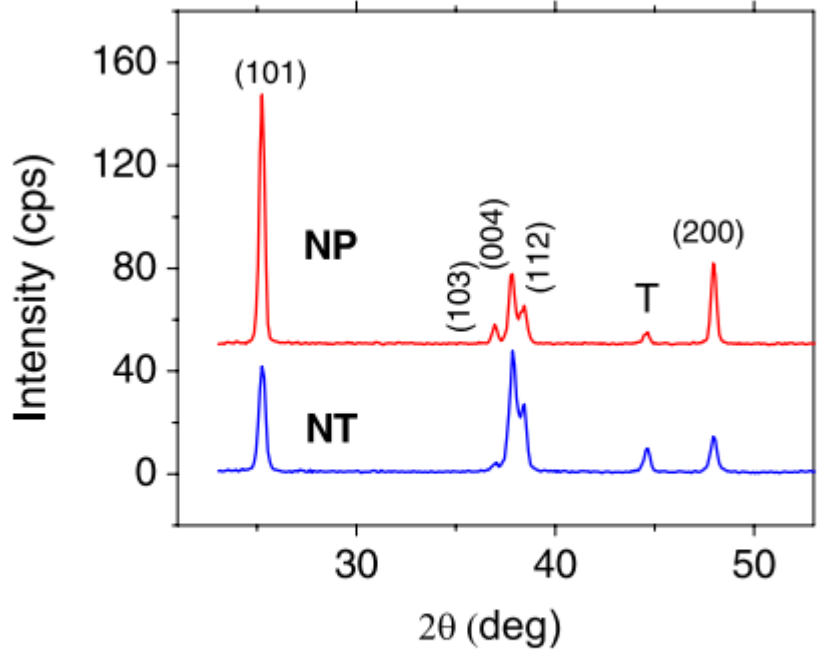
Extended surface area

Next, new research

using this newly NT-based NP

16 Nanotube-based newly TiO₂ NP(part three)

Electronic Structure of TiO₂ Nanotube-based Photocatalysts



(101)(103)(002) anatase peaks intensity for the NP side is 2.5 times greater than the NT side

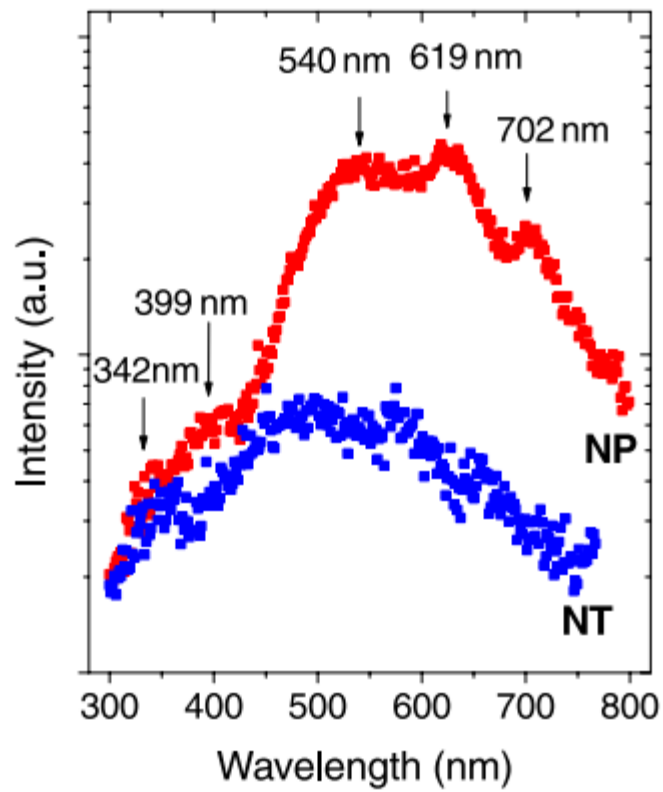
the full width at half-maximum of these three peaks of the NP side is smaller than the NT side

Crystal quality: NP>NT

GAXRD spectra of the NP side (labeled 'NP') and the NT side (labeled 'NT') of the TiO₂ nanostructure sample. T refers to the diffraction peak of Ti foil.

(004)(112) peaks intensity for the NP side is 1.7 times lower than the NT side

More preferential orientation of TiO₂ NTs compared to NPs



Room temperature photoluminescence spectra of the NP side (labeled 'NP') and the NT side (labeled 'NT') of the TiO₂ sample.

Conventional anatase peaks:

- 342nm(3.62eV):** band-to-band direct recombination in TiO₂
- 540nm(2.3eV):** radiative recombination of self-trapped excitons localized on TiO₆ octahedra

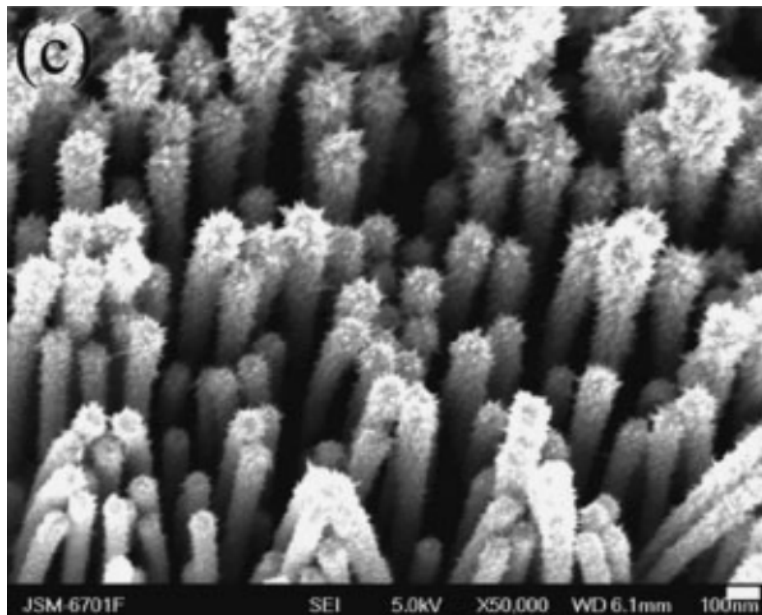
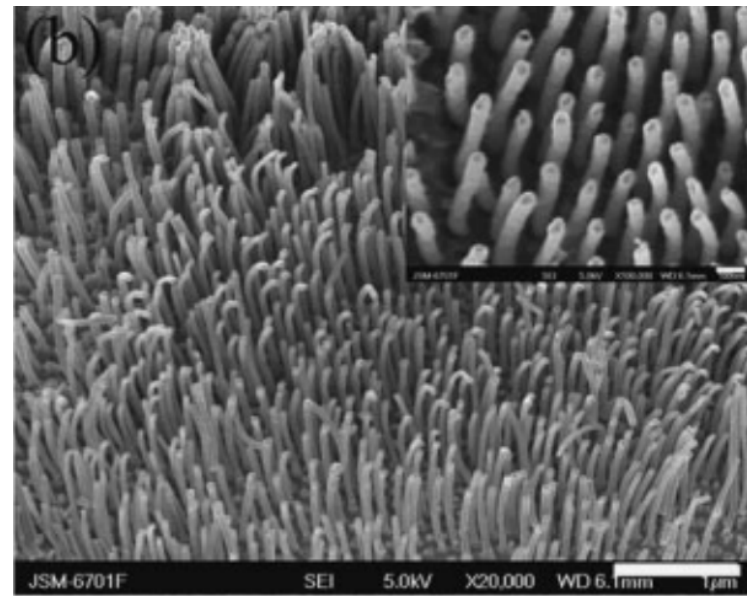
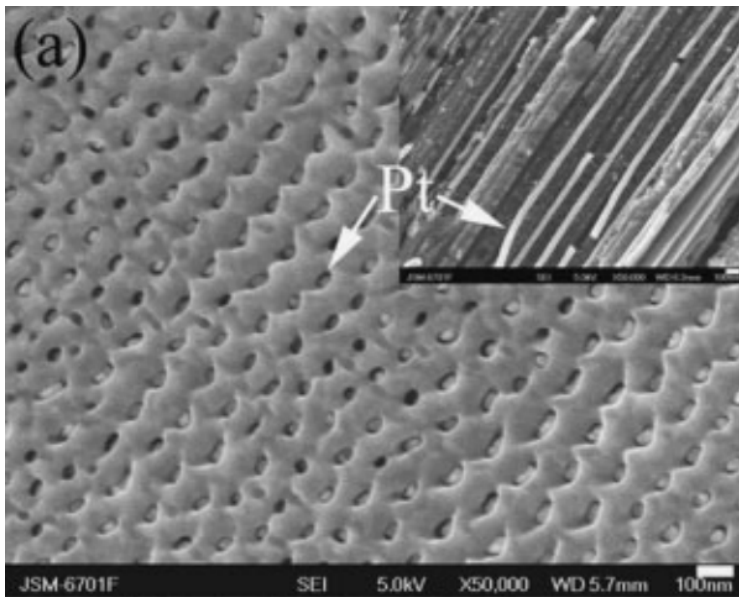
Newly additional peaks for NP:

- 399nm(3.1eV):** radiative annihilation of excitons
- 619nm(2.0eV) and 702nm(1.76eV):** defect-mediated electron-hole recombination

The detailed mechanism of the luminescence process in this newly structure is still remained to be well understood. However, for photoluminescence technique, only the electronic transitions from valence electrons to the conduction band are excited

18 Nanowire growth based on TiO₂ NTs(part four)

Electronic Structure of TiO₂ Nanotube-based Photocatalysts



The composite materials of TiO₂ NTs and the 1D deposited Nanowire can be used as newly photocatalyst

D Wang, B Yu et al. Advanced Materials, 2009, 1964-1967

Research continues.....
Thank you